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# Low-Dimensional Structure and Magnetism of the Quantum Antiferromagnet Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> and the Structure of Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub>

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**Abstract:** Single crystals of the quantum low-dimensional antiferromagnet  $Rb_4Cu(MoO_4)_3$  and the nonmagnetic analogue  $Ru_4Zn(MoO_4)_3$  have been synthesized by a flux-growth method. Detailed structural studies indicate that the Cu(II)—O network separated by a  $MoO_4$  layer has a strongly anisotropic hybridization along the *a*-axis, forming a quasi-one-dimensional (1-d) chain of Cu(II) S = 1/2 spins. Furthermore, our low-temperature thermodynamic measurements have revealed that a quantum paramagnetic state with Wilson ratio ~2 remains stable down to at least 0.1 K, 100 times lower than the intrachain antiferromagnetic coupling scale. The low-temperature magnetic and thermal properties are found to be consistent with theoretical predictions made for a 1-d network of S = 1/2 spins.

### Introduction

Low-dimensional spin systems have received considerable attention because of the possible emergence of novel magnetism such as quantum critical phenomena,<sup>1</sup> Haldane gap formation,<sup>2–4</sup> and field-induced Bose—Einstein condensation.<sup>5</sup> Quantum fluctuations enhanced by low dimensionality may suppress magnetic ordering and give rise to a variety of exotic phenomena. In order to understand the role of quantum fluctuations in low-dimensional spin systems, it is important to explore its phenomenology in simple and well-controlled model systems.

Oxides containing Cu(II) ions have provided a number of model systems of low-dimensional magnets with S = 1/2. The quasi-two-dimensional (2-d) La<sub>2</sub>CuO<sub>4</sub>,<sup>6</sup> the parent compound of the high- $T_c$  cuprates, is a quantum Heisenberg antiferromagnet on a square lattice, and SrCu<sub>2</sub>(BO<sub>3</sub>)<sub>2</sub><sup>7-9</sup> is a quantum dimer system on a Shastry–Shoutherland lattice.<sup>10</sup> One-dimensional

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spin chain systems in copper oxides also provide interesting ground states. For example, an antiferromagnetic Heisenberg model with S = 1/2 spins is theoretically expected to remain disordered down to T = 0 K without forming any long-range order owing to the quantum fluctuations. Instead, a quantum critical state should appear because of divergent spin-spin correlation length as temperature goes to zero. In this quantum spin liquid state, an elementary excitation called spinon emerges, which may be described by a traveling domain wall.<sup>11</sup> Despite such theoretical understanding, the model compounds of such quasi-1-d magnets are still scarce, and thus the study of the quantum critical phase is important. In addition, such 1-d chain magnets hitherto investigated experimentally often exhibit longrange ordering due to finite interchain interactions. Another example,  $Sr_2CuO_3$ , is one of the most studied quasi-1-d S =1/2 antiferromagnets, and in spite of the large exchange coupling  $J \approx 1300$  K, neutron scattering measurements show that this has three-dimensional long-range ordering below  $T_{\rm N} \approx 5.4$  K, indicating a rather small coupling ratio  $J'/J \approx 10^{-5}$ , where J' is the interchain coupling.<sup>12,13</sup> Another prototypical compound is copper benzoate,  $Cu(C_6H_5COO) \cdot 3H_2O$ , with J = 8.6 K, and for this magnet, very interestingly, no long-range order is observed down to 0.02 K by the muon spin relaxation measurement.<sup>14</sup> Sr<sub>2</sub>Cu(PO<sub>4</sub>)<sub>2</sub> is considered one of the best realizations of a 1-d Heisenberg chain in terms of the small ratio J'/J, and

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first-principle calculations have been employed to make a comparison between experimental results and theoretical models.<sup>15</sup> Several years ago, the synthesis and magnetic characterization of the first structurally perfect copper hydroxide chloride mineral, Zn[Cu<sub>3</sub>(OH)<sub>6</sub>Cl<sub>2</sub>], with spin of S = 1/2 Kagomé antiferromagnet, was realized.<sup>16</sup> More recently, a metal–organic hybrid material, Cu(1,3-bdc) (1,3-bdc =1,3-benzenedicarboxylate), has also been shown to exhibit a novel, structurally perfect S = 1/2 Kagomé antiferromagnet, which has opened up an area for the study of quantum disordered phases.<sup>17</sup>

Molybdate compounds exhibit a wide variety of properties, including catalytic gas-phase oxidation of organic compounds, magnetism, magnetoresistance, and ferroelectricity. When they contain a MoO<sub>4</sub> tetrahedral unit, 4d transition metal Mo<sup>6+</sup> ions form a closed shell with S = 0. Thus, a corner-sharing MoO<sub>4</sub> layer may serve as a nonmagnetic block to separate magnetic ions and to construct a low-dimensional network of spins. A well-known example is RbFe(MoO<sub>4</sub>)<sub>2</sub>,<sup>18</sup> a model system of a quasi-2-d triangular antiferromagnet with S = 5/2. It shows a variety of interesting phenomena, such as low-temperature multiferroic properties and a 1/3 magnetic plateau phase with a magnetic field. In the plateau phase, an increase of magnetization with applied magnetic field is suppressed due to its characteristic spin structure, such as an "up-up-down" state.<sup>19</sup>

In our search for low-dimensional quantum antiferromagnets (S = 1/2), we have succeeded in growing high-quality single crystals of Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> and Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub>. In previous literature reports, three types of the temperature-dependent structures were found, indicating that Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> undergoes two temperature-dependent structural phase transitions.<sup>20</sup> The structure was reported to adopt a hexagonal cell (a = 6.15 Å, c = 23.23 Å) above room temperature, an orthorhombic phase (a = 6.074 Å, b = 21.20 Å, c = 23.21 Å) at room temperature, and a monoclinic cell (a = 5.999 Å, b = 21.08 Å, c = 23.03Å;  $\gamma = 90.06^{\circ}$ ) around 103 K. A similar orthorhombic lattice was also reported for  $\beta$ -Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub>, which adopts the *Pn*2<sub>1</sub>*a* space group with cell dimensions of a = 10.90 Å, b = 22.42Å, and c = 6.271 Å; however, no other structural information was reported for this compound.<sup>20</sup> In this article, we report the synthesis, full structure determination, and physical properties of the quantum antiferromagnet Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> and its nonmagnetic reference material Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub>. We also report the magnetic and thermal properties of Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> as a new candidate quasi-1-d Heisenberg quantum antiferromagnet.

#### **Experimental Section**

**Synthesis.** The polycrystalline samples of  $Rb_4Cu(MoO_4)_3$  and  $Rb_4Zn(MoO_4)_3$  are synthesized via ceramic reaction method according to the chemical reaction,

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$$2Rb_{2}CO_{3} + MO + 3MoO_{3} \rightarrow Rb_{4}M(MoO_{4})_{3} + 2CO_{2} \uparrow (M = Cu, Zn)$$

Here, because Rb<sub>2</sub>CO<sub>3</sub> is highly hygroscopic, it was necessary to dehydrate by heating at 300 °C for over 3 h before grinding. The mixture was pressed into a pellet and calcined at 480 °C for 40 h. This procedure was repeated three times until a single phase was obtained. The single crystals were grown by using Rb<sub>2</sub>Mo<sub>2</sub>O<sub>7</sub> as flux. We used the polycrystalline sample as a charge, and the charge—flux molar ratio was 1:1. After the melt solutions were kept at oxygen atmosphere in a fused-silica tube at 590–600 °C for 40 h, they were cooled to 500 °C at 1 deg/h. Greenish transparent hexagonal cylindrical Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> crystals and colorless transparent thin crystals of Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub> were obtained. There were no impurities in the samples grown, as confirmed by X-ray powder diffraction.

Structure Determination. The single-crystal X-ray diffraction data of Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> and Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub> were collected at 298 and 100 K, respectively. Due to the extremely hygroscopic nature of the samples, the crystals were placed in Paratone-N oil to protect the sample from exposure to air and moisture. A crystal with approximate dimensions of  $0.05 \times 0.10 \times 0.125$  mm<sup>3</sup> was mounted onto a glass fiber of the goniometer and placed on a Nonius Kappa CCD X-ray diffractometer (Mo K $\alpha$  = 0.71073 Å). Temperature was regulated with a cooled nitrogen gas stream produced by an Oxford cryostream cooler. The unit cell parameters were determined from images taken at a rotation of  $15^{\circ}\varphi$ . Initial structural models were solved by SIR97 and refined by direct methods against  $F^2$  by full-matrix least-squares techniques using SHELXL97.<sup>21</sup> The data were corrected for absorption, and the displacement parameters were refined as anisotropic. Structural refinement data for Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> and Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub> are given in Table 1, and the atomic parameters are given in Tables 2 and 3 for Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> and Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub>, respectively. The lattice dimensions of Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> are similar to those of the orthorhombic compound  $K_4Zn(MoO_4)_3$ , which crystallizes in the space group  $P2_12_12_1$ .<sup>22,23</sup> An attempt was made to refine the structural model of Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> using the structural information from the  $K_4Zn(MoO_4)_3$ , and although an acceptable R-value was obtained, the atomic displacement parameters for

Table :	1.	Crysta	llographic	Data	for	Rb <sub>4</sub> Cu(MoO <sub>4</sub> ) <sub>3</sub>	and
Rb₄Zn(	M	$0O_4)_3$					

	Crystal Data	
formula	Rb <sub>4</sub> Cu(MoO <sub>4</sub> ) <sub>3</sub>	Rb <sub>4</sub> Zn(MoO <sub>4</sub> ) <sub>3</sub>
space group	Pnma	Pbca
a (Å)	10.581(2)	12.485(2)
<i>b</i> (Å)	23.213(4)	10.8750(10)
<i>c</i> (Å)	6.078(1)	22.2660(10)
$V(Å^3)$	1492.9(5)	3023.2(6)
Ζ	4	8
crystal dimens (mm <sup>3</sup> )	$0.075 \times 0.15 \times 0.175$	$0.05\times0.10\times0.125$
temperature (K)	298	100
density (g cm <sup>-3</sup> )	3.939	3.898
$\theta$ range (°)	3.46-30.02	2.45-30.02
$\mu \text{ (mm}^{-1})$	16.872	16.843
Data	Collection and Refineme	nt
collected reflns	3948	8235
unique reflns	1582	2922
R <sub>int</sub>	0.0295	0.0535
h	$-14 \le h \le 14$	$-17 \le h \le 17$
k	$-32 \leq k \leq 32$	$-15 \le k \le 15$
1	$-8 \leq l \leq 8$	$-31 \leq l \leq 31$
$\Delta \rho_{\rm max}$ (e Å <sup>-3</sup> )	2.333	2.495
$\Delta \rho_{\rm min}$ (e Å <sup>-3</sup> )	-1.226	-1.940
GoF	1.052	1.03
extinction coefficient	0.00084(11)	0.000134(18)
$R_1(F)$ for $F_0^2 > 2\sigma(F_0^2)^a$	0.0508	0.0456
$R_{\rm w}(F_{\rm o}^{2})^{b}$	0.1103	0. 0919
$\frac{{}^{a} R_{1}(F) = \sum   F_{o}  - \sum   F_{o}  - \sum   F_{o} $	$ F_{\rm c} /\Sigma F_{\rm o} .$ <sup>b</sup> $R_{\rm w}(F_{\rm o}^{2})$ =	$= \sum [w(F_{\rm o}^2 - F_{\rm c}^2)^2]/$

Table 2. Atomic Coordinates and Anisotropic Displacement Parameters for Rb<sub>4</sub>Cu(MoO<sub>4</sub>)

				· ,		
atom	Wyckoff site	X	у	Z	U <sub>eq</sub> (Ų) <sup>a</sup>	occupancy
Rb1	8 <i>d</i>	0.33338(8)	0.47125(3)	0.50086(12)	0.0258(2)	1
Rb2	8d	0.33349(10)	0.66056(3)	0.47298(16)	0.0378(2)	1
Mo1	8d	0.00050(6)	0.59262(2)	0.50809(10)	0.01809(17)	1
Mo2	8d	0.16952(16)	0.73000(6)	0.9616(3)	0.0285(2)	0.5
Cu1	4c	0.00041(14)	1/4	0.5088(2)	0.0223(3)	1
O1	8d	0.4181(6)	0.5647(3)	0.2159(10)	0.0361(15)	1
O2	4c	0.4590(12)	1/4	0.603(3)	0.075(4)	1
O3	8d	0.4280(7)	0.5690(3)	0.7497(11)	0.0410(16)	1
O4	8d	-0.0003(8)	0.6692(3)	0.4940(14)	0.054(2)	1
O5	8d	0.1542(6)	0.5672(3)	0.5002(11)	0.0388(15)	1
O6	4c	0.8410(11)	1/4	0.325(2)	0.057(3)	1
07	4c	0.6865(11)	1/4	0.927(2)	0.070(4)	1
O8	8d	0.1736(13)	0.6559(5)	0.983(2)	0.044(4)	0.5

<sup>*a*</sup>  $U_{eq}$  is defined as one-third of the trace of the orthogonalized  $U_{ij}$  tensor.

Table 3. Atomic Positions and Displacement Parameters for  $Rb_4Zn(MoO_4)_3$ 

atom	Wyckoff site	x	у	Ζ	$U_{\rm eq}~({\rm \AA}^2)^a$
Rb1	8c	0.62709(5)	0.91860 (6)	0.97183(3)	0.01378(15)
Rb2	8c	0.87567(5)	0.08440(6)	0.03151(3)	0.01385(15)
Rb3	8c	0.10545(6)	0.57003(6)	0.83948(3)	0.01477(15)
Rb4	8c	0.62453(5)	0.59903(6)	0.84265(3)	0.01486(16)
Zn1	8c	0.32067(7)	0.71724(7)	0.75429(3)	0.01259(18)
Mo1	8c	0.37662(4)	0.74717(5)	0.90582(2)	0.00993(13)
Mo2	8c	0.87294(4)	0.75407(5)	0.90704(2)	0.01007(13)
Mo3	8c	0.37491(5)	0.42223(5)	0.73384(3)	0.01156(14)
O1	8c	0.3690(4)	0.7428(4)	0.6740(2)	0.0205(11)
O2	8c	0.4986(4)	0.4614(4)	0.7629(2)	0.0165(10)
O3	8c	0.3980(4)	0.7525(4)	0.82513(19)	0.0167(11)
O4	8c	0.5121(4)	0.1802(4)	0.9333(2)	0.0188 (11)
O5	8c	0.2808(4)	0.5400(4)	0.75474(19)	0.0168(10)
06	8c	0.1704(4)	0.7760(4)	0.7622(2)	0.0178(10)
07	8c	0.2594(4)	0.6681(5)	0.9229(2)	0.0224(12)
08	8c	0.3683(4)	0.8969(4)	0.9333(2)	0.0201(11)
09	8c	0.3833(4)	0.4173(4)	0.65647(19)	0.0195(11)
O10	8c	0.4851(4)	0.6700(4)	0.9381(2)	0.0206(11)
O11	8c	0.6227(4)	0.4084(4)	0.92981(19)	0.0167(11)
012	8 <i>c</i>	0.7585(4)	0.6840(4)	0.93782(19)	0.0174(11)

 $^{a}U_{\rm eq}$  is defined as one-third of the trace of the orthogonalized  $U_{ij}$  tensor.

multiple atoms were large and not well-behaved. Also, the intensities of the observed structure factors were generally higher than the calculated values, indicating either a twinned or incorrect absolute structure. These differences all contribute to choosing the higher symmetry *Pnma* model over the published  $P2_12_12_1$  model, and this is not unexpected, given that the difference between the two structure types is the transition metal coordination environment, where the Zn in K<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub> adopts a tetrahedral bonding environment while the Cu in Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> resides in a distorted square planar environment.

**Magnetic and Thermal Properties.** We measured the magnetization M down to 1.9 K under a magnetic field  $\mu_0 H$  up to 7 T with a MPMS SQUID magnetometer to investigate the magnetic properties of Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub>. The specific heat C was measured by the thermal relaxation method using a He<sup>3</sup> and/or dilution refrigerator down to 0.1 K for Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> and using a He<sup>3</sup> refrigerator down to 0.4 K for Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub>.

#### **Results and Discussion**

**Crystal Structure of Rb**<sub>4</sub>**Cu**(MoO<sub>4</sub>)<sub>3</sub>. Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> adopts the *Pnma* space group with lattice dimensions of a = 10.581(2)

Å, b = 23.213(4) Å, and c = 6.078(1) Å, and selected bond distances and angles are provided as Supporting Information. Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> is shown in Figure 1, and the structure consists of distorted square planar CuO<sub>4</sub> bonded to two slightly distorted MoO<sub>4</sub> tetrahedral units in the *c*-direction. Mo1 adopts a slightly distorted tetrahedral environment, with interatomic distances ranging from 1.731(6) to 1.780(6) Å. Mo2 is also slightly distorted in a tetrahedral environment, with interatomic distances ranging from 1.674(13) to 1.805(12) Å. Mo2 is disordered, with an occupancy of 0.5. Rb1 is coordinated to 10 oxygen atoms, with bond distances ranging from 2.905(6) to 3.294(7) Å. Rb2 is coordinated to six oxygen atoms, with bond distances ranging from 2.862(6) to 3.205(11) Å.

Figure 2 shows the 1-d Cu chains with Cu–O distances ranging from 1.876(6) to 2.023(11) Å and O-Cu-O angles of 157.6(5)° and 179.0(5)°. Cu superexchange is possible because the layers of Cu subunits are connected by the MoO<sub>4</sub> subunits. There are two types of superexchange paths, J and J', as shown in Figure 2a. The distorted square pyramid CuO<sub>5</sub> environment should stabilize a lone pair in the  $d_{x^2-y^2}$  orbital, as also seen in other oxide structures such as La<sub>2</sub>CuO<sub>4</sub>.<sup>6</sup> Along the superexchange path J, the disordered, slightly distorted MoO<sub>4</sub> tetrahedral units link the CuO<sub>5</sub> subunits. This path includes the strong  $\sigma$ bond between the  $d_{x^2-y^2}$  orbital of Cu<sup>2+</sup> and the  $p_{x,y}$  orbitals of O<sup>2-</sup>. On the other hand, the main magnetic link along the superexchange path J' must involve the coupling between the  $Cu^{2+}$  ion and the  $O^{2-}$  ion at the apical site of the square pyramid CuO<sub>5</sub>. This coupling should be weak because of (1) the long Cu-O distance of 2.4 Å and (2) the almost orthogonal spatial configuration for the relation between the  $d_{x^2-y^2}$  orbital of Cu<sup>2+</sup> and the  $p_z$  orbital of  $O^{2-}$ , which results in a negligibly small superexchange bond J' in comparison with J. To illustrate this, we draw the local coordination and the  $Cu^{2+} d_{x^2-y^2}$  and the  $O^{2-}$  $p_z$  orbitals as shown in Figure 2b. Thus, one localized electron in the anisotropic  $d_{x^2-y^2}$  orbital is likely to form a strong superexchange path J along the a-axis and organize a 1-d spin chain. This 1-d spin coupling is counterintuitive, because the material has an apparently layered structure where a Cu<sup>2+</sup> ion is densely distributed to form a distorted triangular lattice in a 2-d layer on the *ac* plane, separated by  $Rb^+$  and  $MoO_4$  block layers. In this system, however, the arrangement of the anisotropic hybridization between Cu 3d and O 2p orbitals stabilizes the 1-d chain of Cu-O-Mo-O-Cu along the *a*-axis, as shown in Figure 2a. Although this orthorhombic structure is similar to that of other molybdates, it actually differs significantly in the geometry of the transition metal atom, as discussed previously.

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*Figure 1.* (a) Crystal structure of  $Rb_4Cu(MoO_4)_3$  as viewed down the *c*-axis, where the blue, green, purple, and red spheres refer to Rb, Cu, Mo, and O atoms, respectively. (b) Distorted Cu square planar environment of  $Rb_4Cu(MoO_4)_3$ .



*Figure 2.* (a) View of the Cu 1-d chains and their bonding environment. Both *J* and *J'* indicate two types of near-neighbor superexchange couplings between S = 1/2 spins at the Cu<sup>2+</sup> ion, connected through Cu–O–Mo–O–Cu paths. The longer Cu–O2 apical contacts are not shown for clarity. (b) The local coordination of CuO<sub>5</sub>, showing the  $d_{x^2-y^2}$  orbital of Cu<sup>2+</sup> and the  $p_x$ ,  $p_y$ , and  $p_z$  orbitals of O<sup>2-</sup>. For simplicity, we show d and p orbitals for the ideal CuO<sub>4</sub> square coordination, neglecting small local distortions.

**Crystal Structure of Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub>.** The structure of Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub> crystallizes in the orthorhombic space group *Pbca* (No. 61) and adopts a new structure type with lattice parameters of a = 12.485(2) Å, b = 10.8750(10) Å, and c = 22.2660(10) Å, as shown in Figure 3. Selected bond distances and angles are provided as Supporting Information. The structure consists of slightly distorted ZnO<sub>4</sub> tetrahedral units surrounded by four slightly less distorted MoO<sub>4</sub> tetrahedra. The Zn tetrahedra form chains with Mo3 tetrahedra along the *b*-axis and are capped by Mo1 and Mo3 tetrahedra in the *c*-direction. The new Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub> structure type is similar to other double molybdates such as Rb<sub>4</sub>Mn(MoO<sub>4</sub>)<sub>3</sub>;<sup>24</sup> however, one notable difference between the Zn analogue and the hexagonal Rb<sub>4</sub>Mn(MoO<sub>4</sub>)<sub>3</sub> is the coordination environment of the transition metal, where the Mn analogue consists of a Mn trigonal

bipyramid while the Zn analogue contains slightly distorted Zn tetrahedra.<sup>25,26</sup> The structure of Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub> is also reminiscent of the orthorhombic K<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub> ( $P2_12_12_1$  space group).<sup>22,23</sup> Although Zn adopts a tetrahedral environment in both analogues, the orthorhombic K<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub> structure type contains disordered MoO<sub>4</sub> subunits, not observed in the Rb analogue.

The O–Mo–O angles in the tetrahedral subunits range from 107.9(2)° to 110.9(2)° for Mo1, 108.5(2)° to 110.7(2)° for Mo2, and 108.2(2)° to 111.5(2)° for Mo3, indicating a slight deviation from the expected angle of 109.5°. The O–Zn–O angles range from 94.28(19)° to 126.3(2)° for the ZnO<sub>4</sub> tetrahedra, showing more distortion than the MoO<sub>4</sub> units. The distances are consistent with the ionic radii of Rb<sup>+</sup>, Zn<sup>2+</sup>, and Mo<sup>6+</sup> as well as the distances in the known  $K_4Zn(MoO_4)_3$  structure.<sup>22,23,27</sup>

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*Figure 3.* (a) Crystal structure of  $Rb_4Zn(MoO_4)_3$  as viewed down the *a*-axis, where the blue, green, purple, and red spheres refer to Rb, Zn, Mo, and O atoms, respectively. (b) Zn bonding environment of  $Rb_4Zn(MoO_4)_3$ .



**Figure 4.** Temperature dependence of the magnetic susceptibilities of  $Rb_4Cu(MoO_4)_3$  below 40 K. Blue (green) solid and open circles show magnetization  $\mu_0H|ac$  (b) at 7 and 0.01 T, respectively. There is no hysteresis between results obtained by zero-field-cooled and field-cooled measurements. The red (black) dashed line shows the result of the fitting by the theoretical calculation by Eggert, Affleck, and Takahashi<sup>29</sup> using the anisotropic *g*-values and the antiferromagnetic coupling *J* estimated from the Curie—Weiss analyses (see text).

**Physical Properties of Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub>.** The one-dimensional nature structurally expected for the exchange interaction between neighboring Cu<sup>2+</sup> ions with S = 1/2 has been confirmed by comprehensive measurements, including the temperature dependence of the susceptibility and the specific heat and the magnetization process under magnetic fields for Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub>. The temperature dependence of the magnetic susceptibilities at  $\mu_0H = 0.01$  and 7 T for  $\mu_0H||ac$  and  $\mu_0H||b$  have a broad peak at 5 K, which is characteristic of the S = 1/2 1-d spin chain system as we will compare them with theoretical calculation below, and shows no anomaly indicative of a phase transition down to 1.9 K (Figure 4). There is no difference in the zero-field-cooled and field-cooled results at 0.01 T. Data were corrected for the diamagnetism of the core electron in

**Table 4.** Magnetic Effective Moment  $p_{eff}$ , g-Value, Weiss Temperature  $\Theta_W$ , and Uniform Susceptibility  $\chi_0$  in Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub>, Obtained by Curie–Weiss Fitting

Rb <sub>4</sub> Cu(MoO <sub>4</sub> ) <sub>3</sub>	µ₀Hllac	µ₀HIIb	
$p_{\rm eff}$	1.91	1.79	
g	2.21	2.07	
$\Theta_{W}(K)$	-5.01	-5.15	
$\chi_0 \text{ (emu/mol)}$	$2.00 \times 10^{-4}$	$3.18 \times 10^{-4}$	

 $Rb_4Cu(MoO_4)_3$ ,  $\chi_{dia} = -256 \times 10^{-6} \text{ emu/mol.}^{28}$  The difference between two susceptibilities for  $\mu_0 H \| ac$  and  $\mu_0 H \| b$  is attributable to the anisotropy of the g-value and the Van Vleck's paramagnetism. We fitted the  $\chi$  data assuming that  $\chi = \chi_0 + C/(T - C)$  $\Theta_{W}$ ) for 40–300 K and obtained the effective magnetic moment  $p_{\rm eff}$ , g-value, Weiss temperature  $\Theta_{\rm W}$ , and  $\chi_0$  as shown in Table 4. The Weiss temperatures are both negative and indicate the antiferromagnetic interaction between neighboring Cu<sup>2+</sup> ions. The constant term  $\chi_0 \sim 2.0 \times 10^{-4}$  emu/mol for  $\mu_0 H ||ac$  and  $3.2 \times 10^{-4}$  emu/mol for  $\mu_0 H \| b$  are most likely due to the Van Vleck's paramagnetism by the orbital hybridization.  $p_{eff}$  for  $\mu_0 H \| b$  is close to the theoretical value 1.73  $\mu_B$  for S = 1/2. On the other hand,  $p_{\rm eff}$  for  $\mu_0 H || ac$  is larger than 1.73  $\mu_{\rm B}$  and is ascribable to spin-orbit coupling. The exchange interaction can be estimated to be  $J \approx 10.0$  K by using the relation  $\Theta_{\rm w} = -[zs(s)]$ (+ 1)J/3, where z = 2 is the number of nearest-neighbor magnetic ions. Using the above anisotropic g-values and the antiferromagnetic coupling constant J, now we compare our  $\chi(T)$ results with the theoretical calculation of S = 1/2 antiferromagnetic Heisenberg 1-d spin chain system with high accuracy obtained by Eggert, Affleck, and Takahashi in Figure 4.29 The agreement is rather good except for a slight deviation at low temperatures below the peak temperature. The slight positive deviation in comparison with theory may come from weak interchain coupling, as will be discussed below.

Figure 5 shows the magnetization measured at 2 K for  $\mu_0 H || ac$ and  $\mu_0 H || b$ . Magnetization curves have a convex downward

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**Figure 5.** Field dependence of the magnetization of a single crystal of Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> up to 7 T, measured at 2 K for  $\mu_0 H || b$  and  $\mu_0 H || ac$ . Solid and open circles show the up-sweep and down-sweep results, respectively.



*Figure 6.* Temperature dependence of the magnetic specific heat of  $Rb_4Cu(MoO_4)_3$ . Red solid circles are the results at 0 T. Inset: Temperature dependence of the total specific heat of  $Rb_4Cu(MoO_4)_3$  (red) and the lattice contribution to the specific heat of  $Rb_4Zn(MoO_4)_3$  (black). The latter is estimated from the specific heat of the nonmagnetic analogue  $Rb_4Zn(MoO_4)_3$  by obtaining thermal variations of its Debye temperature using the Debye equation.<sup>36</sup>

curvature, as often seen in S = 1/2 systems.<sup>30</sup> Normally, M shows the convex upward dependence of H, obeying the Brillouin function, when there is no magnetic correlation. Meanwhile, if the antiferromagnetic correlation develops, M is linear over a wide range of H. However, in S = 1/2 systems, if the antiferromagnet correlation develops, spin singlet pairs are partially formed, and the average length of spins shortens, owing predominantly to quantum fluctuations. Because the length of spins elongates again with applied field, the slope of the magnetization curve is enhanced by applying field. We have also carried out specific heat measurements for Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> under various fields. Figure 6 shows the temperature dependence of the magnetic specific heat  $C_{\rm M}$  at  $\mu_0 H = 0$  T.  $C_{\rm M}$  was estimated by subtracting the lattice contribution, which was estimated using the total specific heat of the nonmagnetic analogue  $Rb_4Zn(MoO_4)_3$ .  $Rb_4Zn(MoO_4)_3$  does not have completely the same structure, but it is similar and provides the best estimate



**Figure 7.** Temperature dependence of the magnetic specific heat of  $Rb_4Cu(MoO_4)_3$  at various fields on a full logarithmic scale. Red, green, and blue solid symbols are the results for  $\mu_0H||ac$  at 0, 1, and 7 T, respectively.

of the lattice specific heat of Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub>. Thus, while the high-temperature estimate of  $C_{\rm M}$  might have a slight systematic error, the discussion below on the most interesting lowtemperature part below 3 K, where the lattice contribution is relatively small, should be independent of the estimate. No anomaly indicative of long-range order was observed down to 0.1 K, despite the Weiss constant of  $|\Theta_W| \approx 5.0$  K. This is most likely because strong one-dimensionality of the spin structure significantly enhances the spin fluctuations and suppresses the magnetic order, stabilizing the quantum disordered state. The magnetic heat capacity  $C_{\rm M}/T$  forms a broad peak at ~2.5 K, as shown in Figure 6, signaling the development of antiferromagnetic correlation. As has been seen in other Heisenberg antiferromagnetic chain systems,<sup>31</sup>  $C_{\rm M}$  exhibits a linear temperature dependence below 0.8 K under zero magnetic field (Figure 7). At T < 0.8 K,  $C_{\rm M}/T$  remains almost constant at  $\gamma \approx$ 0.64 J/mol·K<sup>2</sup> as  $T \rightarrow 0$ . For 1-d spin chain systems, it is generally expected that the field with the energy scale lower than the peak temperature of  $C_{\rm M}/T$  (in our case,  $\sim 3$  T) does not change the linear temperature dependence of the specific heat, but the one beyond the peak temperature scale opens the Zeeman gap in the spin excitation spectrum and thus suppresses the linear dependence. Indeed, we observe that a magnetic field of 1 T does not change the temperature dependence, but 7 T is enough to suppress the peak temperature down to 1.5 K, and thus the specific heat is no longer linear with temperature under 7 T.

Here, we compare our results with the exact solutions available for thermodynamics of the S = 1/2 antiferromagnetic Heisenberg chain.<sup>32–34</sup> First, the susceptibility is theoretically estimated to be  $x_{\rm th}^{\rm max} J/Ng^2\mu^2 = 0.1469279(1)$  at the peak temperature  $T_{\rm th}^{\rm max} = 0.6408510(4)J$ . Consequently, one can obtain a *J*-independent parameter,  $\chi_{\rm th}^{\rm max}T_{\rm th}^{\rm max} = 0.0353229(3)g^2$  emu/mol. Using experimentally obtained  $g_{\rm ac} = 2.21$ ,  $\chi_{\rm th}^{\rm max}T_{\rm th}^{\rm max}$  is estimated to be 0.1725 emu/mol, which is close to the experimental value  $\chi_{\rm ex}^{\rm max}T_{\rm ex}^{\rm max} = 0.155$  emu/mol at  $\mu_0 H = 0.01$  T for  $\mu_0 H || ac$  (Figure 4). Second, the peak position of the specific

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heat by Padé approximations locates at  $C_{\rm th}^{\rm max}R = 2.9075$ J/mol·K and  $T_{\text{th}}^{\text{max}} = 4.8$  K for  $\mu_0 H = 0$  T, where J = 10.0 K was used and R is the molar gas constant. These are again close to the experimental values for the specific heat as shown in Figure 6, namely  $C_{ex}^{max} = 2.6436 \text{ J/molCu} \cdot \text{K}$  at  $T_{ex}^{max} = 4.0$ K. Finally, the Wilson–Sommerfeld ratio  $R_W$  for the S = 1/2antiferromagnetic Heisenberg chain is given by the equation  $R_{\rm W}(t) = 4\pi^2 \chi^*(t) t/3C(t)$ , where the spin susceptibility  $\chi^*(t)$ , the magnetic specific heat C(t), and t are defined as  $\chi J/Ng^2 \mu_B^2$ , CJ/ $Nk_{\rm B}^2T$ , and  $k_{\rm B}T/J$ , respectively.<sup>29</sup> Theoretically,  $R_{\rm W}$  for the S =1/2 antiferromagnetic Heisenberg chain is expected to be exactly 2 for  $t \rightarrow 0$ . Experimental values of  $R_{\rm W}$  are estimated to be 2.264 for  $\mu_0 H \| ac$  and 2.217 for  $\mu_0 H \| b$  at t = 0.2. These are 10% larger than the theoretical values but roughly consistent with the theory. All these observations indicate that the system can be regarded as a S = 1/2 Heisenberg chain having a quantum spin disordered state at least down to 0.1 K. This is also consistent with the structure, based on the fact that the hole orbital  $d_{x^2-y^2}$  of Cu<sup>2+</sup> in the CuO<sub>4</sub> square is linked along the *a*-axis through the p orbital of  $O^{2-}$  ions. On the other hand, as we discussed above, the magnetic interchain coupling J' in the ac plane should be weak but must still be finite and may contribute to the low-temperature magnetic properties to some extent. Here, J' couples the nearest-neighbor Cu(II) spins in different chains to form an anisotropic triangular lattice antiferromagnet in the ac plane, namely, in the layer consisting of the CuO<sub>5</sub> square pyramid (Figure 2a). For this type of triangular antiferromagnets with spatially anisotropic exchange couplings, a high-temperature series expansion has been made to estimate the variation of the characteristic parameters of the susceptibility curves as a function of J/(J+J'), such as  $T^{\text{max}}/\Theta_{\text{W}}$ ,  $\chi(4T^{\text{max}})/2$  $\chi^{\text{max}}$ , and  $T^{\text{max}}\chi^{\text{max}}/Ag^2$  ( $A = N_{\text{A}}g^2\mu_{\text{B}}^2/4k_{\text{B}} = 0.0938$  in cgs units).<sup>35</sup> Using the  $\chi(T)$  results at 0.1 T (Figure 4), these parameters are found to be 1.05, 0.56, and 0.35 for  $\mu_0 H \| ac$  and 1.02, 0.57, and 0.37 for  $\mu_0 H || ab$ , respectively, thus yielding the estimate of  $J/(J + J') \approx 0.95$ , namely  $J'/J \approx 0.05$ , according to the theory. This suggests that the interchain coupling is actually weak, as expected from the crystal structure. Therefore, Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> can be regarded as a nearly ideal 1-d antiferromagnetic Heisenberg system.

For Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub>, the specific heat measurement was carried out, and its temperature dependence is shown in Figure 8. No structural phase transition is observed down to 1.8 K, consistent with the results of crystal structure analysis. As we mentioned above, we used this system as a reference material to obtain the lattice contribution to the total specific heat in Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub>. Our detailed structural studies of a quantum low-dimensional magnet, Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub>, indicate that a Cu(II) network separated by a MoO<sub>4</sub> layer forms a 1-d chain of S =1/2 spin. Furthermore, we have found a quantum spin liquid-



Figure 8. Temperature dependence of the total specific heat of  $Rb_4Zn(MoO_4)_3$ .

like feature at low temperature above 0.1 K, consistent with 1-d network of S = 1/2 spin.

## Conclusion

In this article, we report a new one-dimensional quantum antiferromagnet, Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub>, and its nonmagnetic reference, Rb<sub>4</sub>Zn(MoO<sub>4</sub>)<sub>3</sub>. Our structural analysis using single crystals of Rb<sub>4</sub>Cu(MoO<sub>4</sub>)<sub>3</sub> strikingly clarifies that the layered compound consists of 1-d chains of  $Cu^{2+} S = 1/2$  on each layer because of the anisotropic hybridization between Cu 3d and O 2p orbitals. The low-temperature physical properties are fully consistent with the theoretical expectation for a quantum spin chain system. This indicates that the strong quantum effect enhanced by the low dimensionality stabilizes a quantum paramagnetic state below  $\sim 1$  K, at least down to 0.1 K, which is 100 times smaller than the nearest-neighbor coupling J. This material may well serve as a model system to deepen our understanding of quantum critical magnetism expected in 1-d spin chains (such as spin-on excitations) through future experiments.

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**Supporting Information Available:** Selected bond distances and angles and additional crystallographic information, including crystallographic information files (CIF). This material is available free of charge via the Internet at http://pubs.acs.org.

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